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At the very center of an atom is the nucleus, which is made up of small particles called protons and neutrons. Protons are very small, positively-charged particles, and neutrons are neutral particles that have no charge. Atoms can have just one proton or they can have multiple.

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The fundamental building block of matter is the atom. An atom consists of three main parts: protons, neutrons, and electrons. Protons have a positive electrical charge. Neutrons have no electrical charge. Electrons have a negative electrical charge. Protons and neutrons are found together in what is called the nucleus of the atom.

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Following are the various models explaining structure of an atom:

1. Thomson's Model of an atom It is also known as Thomson's plum pudding model.

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