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Solutions- Provide a statement of the solutions (what does this mean answer mean and how does it relate to concepts) Chemistry 12 work on the calculations of enthalpies, bond energies and hess' law from the study guide. If you have finished you can start reading on Chapter 5.6 Present and Future and Energy Sources p. 325

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Pre-laboratory Assignment: Calorimetry and Hess's Law Show your work for all calculations. When 1.104 grams of iron metal are mixed with 26.023 grams of hydrochloric acid in a coffee cup calorimeter, the temperature rises from 25.2 °C to a maximum of 33.5 °C.

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• laboratory safety rules • scientific problem solving • graphing, and solving linear and exponential equations You can review prerequisite concepts and skills on the Nelson Web site, in the Chemistry Review unit, and in the Appendices. A Unit Pre-Test is also available online. Prerequisites matter pure B substances C elements homogeneous ...

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