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- Part 1 شاطر 6 Calorimetry Examples:

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 of our very existence as individuals and
 as a society. The food that we eat
 furnishes the energy to live, work, and
 play, just as the coal and oil consumed

by manufacturing and transportation
 systems power our modern industrialized
 civilization. In the past, huge quantities
 of carbon-based fossil fuels have been
 available for the taking.Chapters 6:
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 Enthalpy is a thermodynamic quantity
 that accounts for heat flow during the

course of a chemical reaction Equals the energy contained by the system as well as the pressure/volume work done on or by the system: $H = E +$

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The work done in gas expansion is equal to the product of the external, opposing pressure and the change in volume. $w = -P\Delta V$ We assume that the volume of liquid water is zero compared to that of steam.CHAPTER 6 THERMOCHEMISTRY - Knowledge DirectorySection 6.2.

Enthalpy and Calorimetry. A 100.0 g sample of water at 90 °C is added to a 100.0 g sample of water at 10 °C. The final temperature of the water is: a) Between 50 °C and 90 °C b) 50 °C c) Between 10 °C and 50 °C.Chapter 6 Thermochemistry - Wunder Chem -

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Chapter 6: Thermochemistry. Energy is

the essence of our very existence as individuals and as a society. The food that we eat furnishes the energy to live, work, and play, just as the coal and oil consumed by manufacturing and transportation systems power our modern industrialized civilization. In the past, huge quantities of carbon-based fossil fuels have been available for the taking.

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